

# General Principles for Hydrocarbon Vapor Intrusion



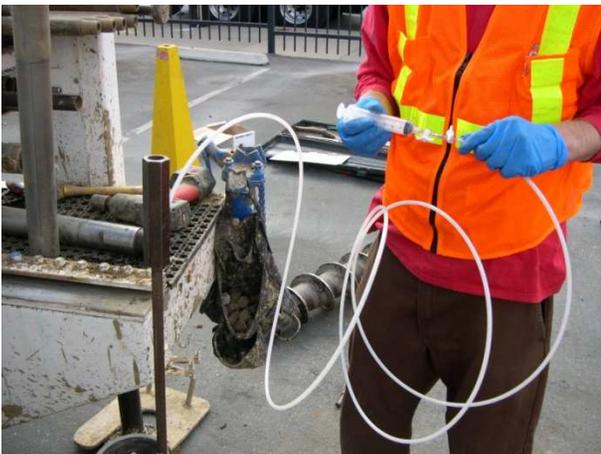
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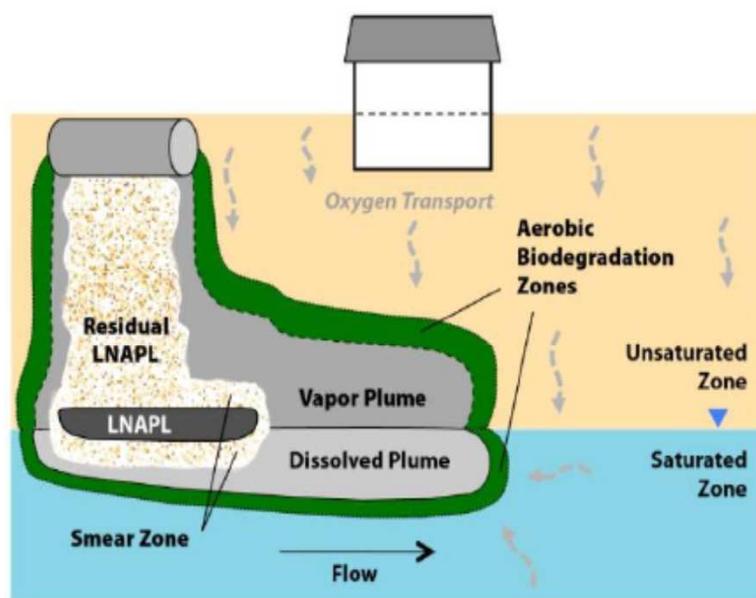
Michigan PVI Workshop

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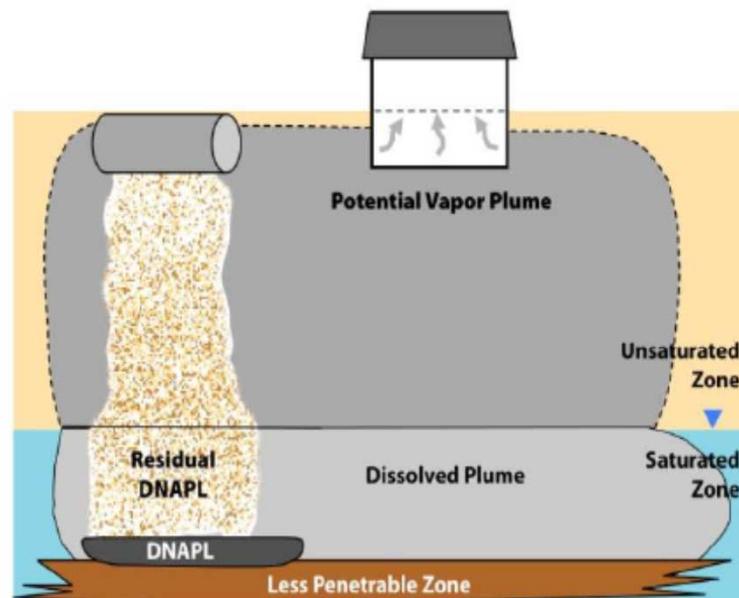


## Petroleum Hydrocarbons And Chlorinated Hydrocarbons Differ In Their Potential For Vapor Intrusion



**Figure 1. Typical petroleum hydrocarbon transport conceptual scenario**

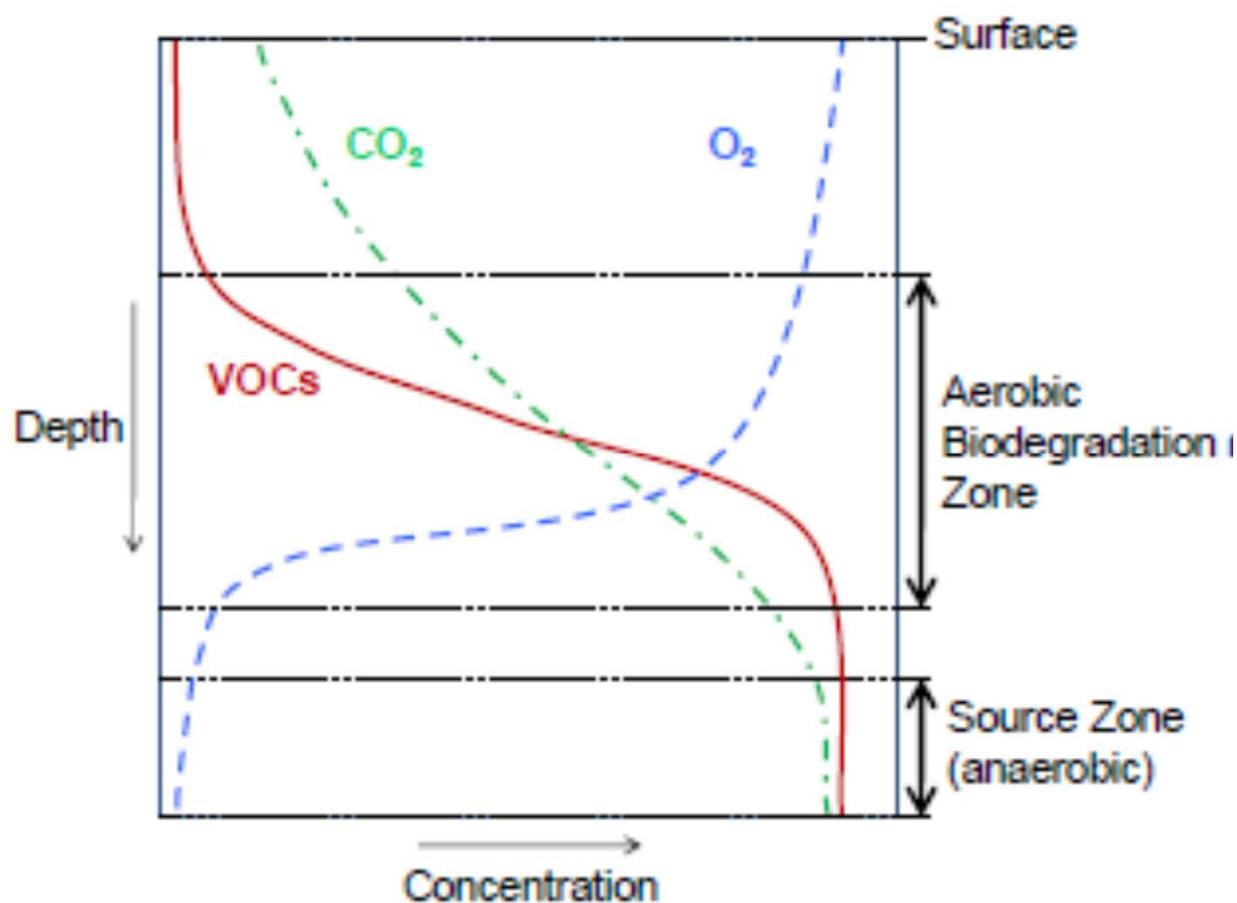
Aerobic biodegradation of PHCs along the perimeter of the vapor and dissolved plumes limits subsurface contaminant spreading. Effective oxygen transport (dashed arrows) maintains aerobic conditions in the biodegradation zone. Petroleum LNAPL (light nonaqueous phase liquid) collects at the groundwater surface (the water table, blue triangle).



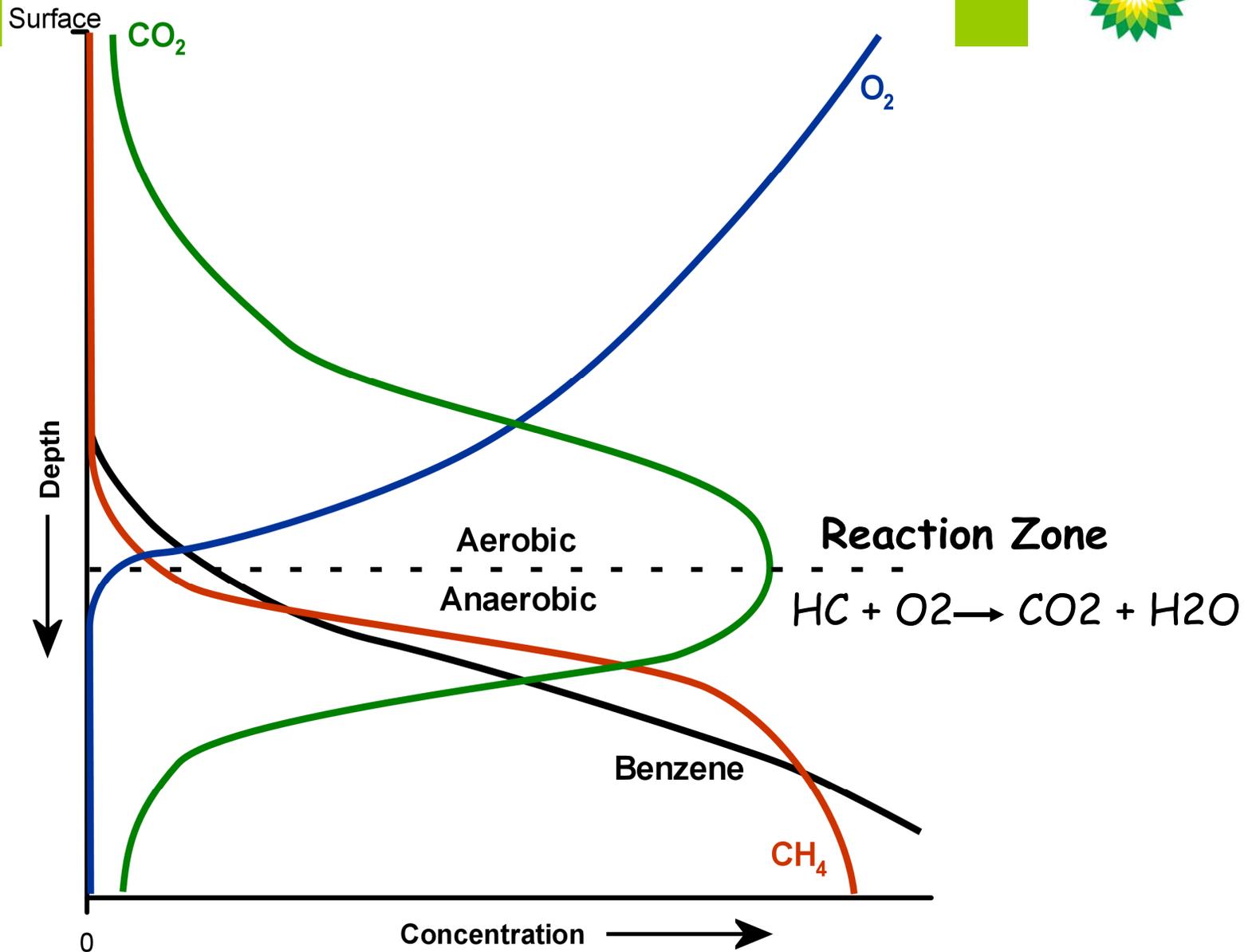
**Figure 2. Typical chlorinated solvent transport conceptual scenario**

Biodegradation of CHCs is anaerobic and usually slower than PHC biodegradation, so that the vapor and dissolved plumes often migrate farther than PHC plumes. CHC DNAPL (dense nonaqueous-phase liquid), if present, can sink below the water table, collecting in this case on a less penetrable layer.

# Typical vertical concentration profile in the unsaturated zone for VOCs, CO<sub>2</sub>, and O<sub>2</sub>



# Clean Soil Model for HC Vapors



# Clean Soils vs Dirty Soils



## CLEAN SOILS

- Have oxygen concentrations above 3%
- Have no residual petroleum hydrocarbons
- PID values will be below 100 ppm
- Usually have low methane and carbon dioxide values
- Non-detect or very low petroleum hydrocarbon vapors



## DIRTY SOILS

Have low oxygen concentrations (at or near 0%)  
Can have residual petroleum hydrocarbons  
PID reading at or above 100 ppm  
Usually have high methane and carbon dioxide values  
Elevated petroleum hydrocarbon vapors



# Importance of Biodegradation of Petroleum Hydrocarbons



Ground Water  
Monitoring & Remediation

## Evidence for Instantaneous Oxygen-Limited Biodegradation of Petroleum Hydrocarbon Vapors in the Subsurface

*by G.B. Davis, B.M. Patterson, and M.G. Trefry*

Ground Water Monitoring & Remediation 29, no. 1 / Winter 2009/pages 126-137

# Oxygen versus Petroleum Hydrocarbon Plots

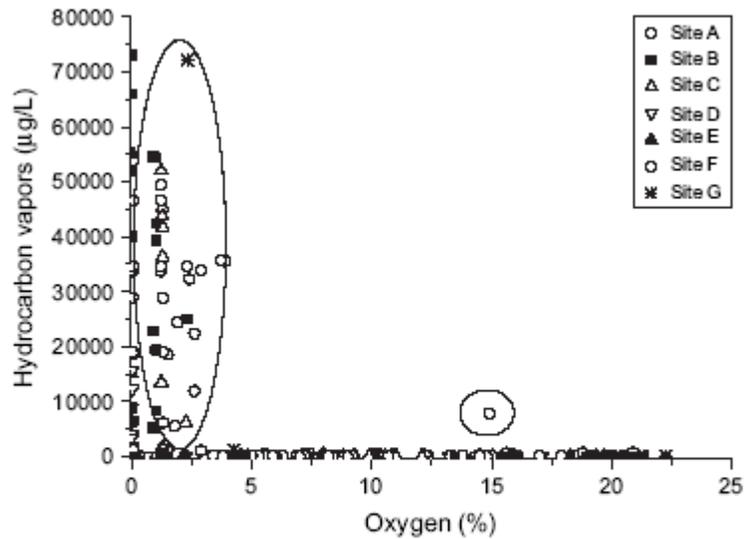


Figure 2. Total petroleum hydrocarbon vapor concentrations compared to oxygen concentrations for Sites A-G. Circled are data from sites where at the same sampling port both oxygen and hydrocarbon vapors were above detection levels.

**Manual collected O2 data:  
Leakage of O2 into syringe**

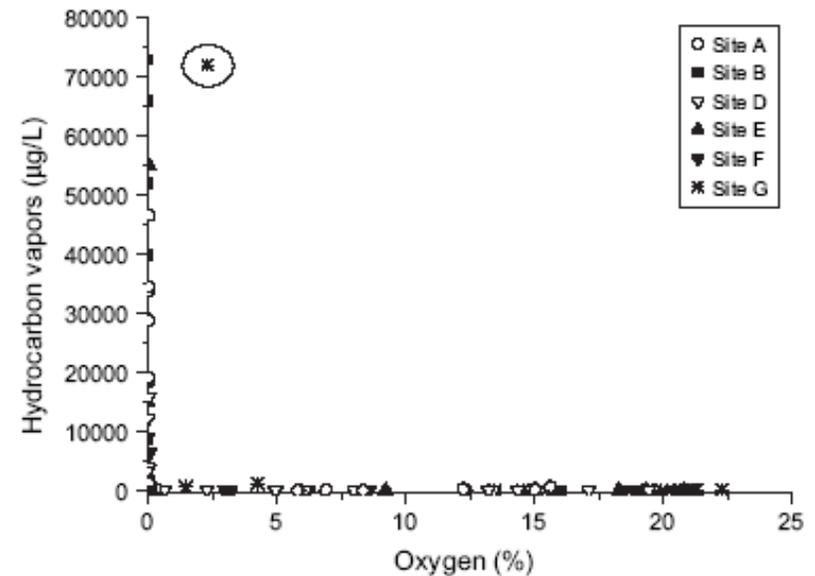


Figure 4. Total petroleum hydrocarbon vapor concentrations compared to oxygen concentrations; data from in situ oxygen and VOC probes for all available sites.

**In situ O2 data**

From Davis et al., 2009

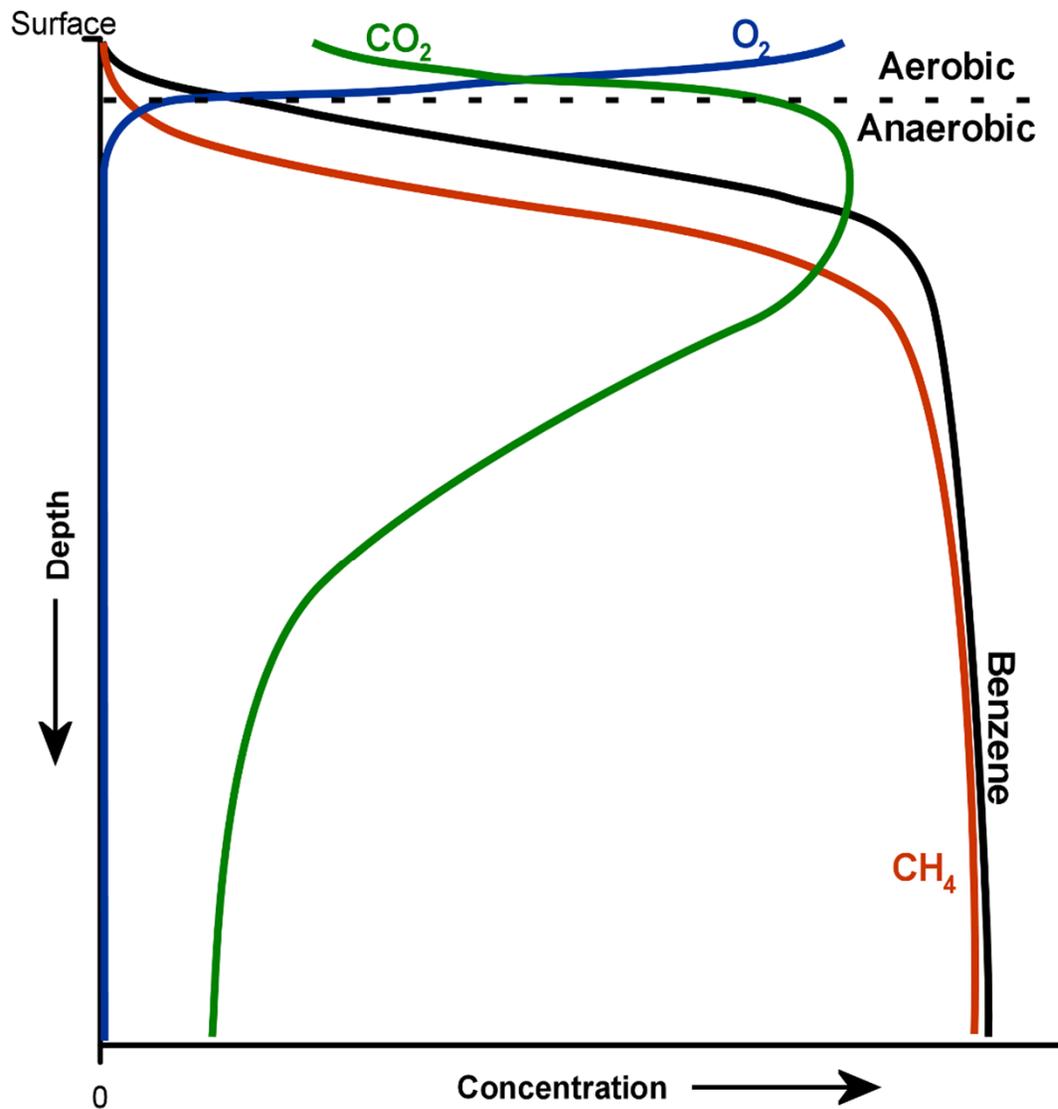
# Is There Enough Oxygen?



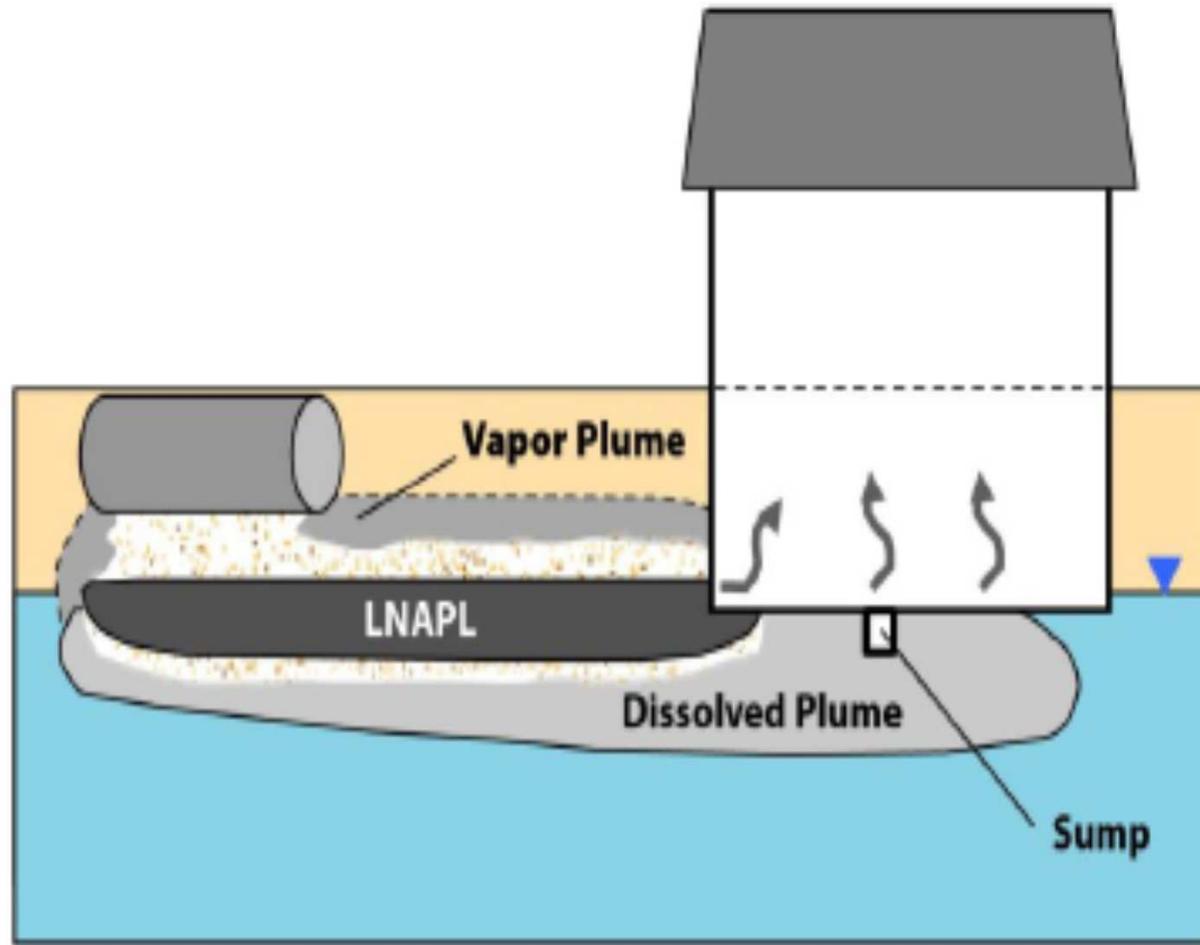
- **Aerobic Biodegradation**
  - Hydrocarbon to Oxygen use ratio: 1 : 3 (kg/kg)
  - Atmospheric air (21% Oxygen; 275 g/m<sup>3</sup> oxygen) provides the capacity to degrade 92 g/m<sup>3</sup> hydrocarbon vapors (92,000,000 ug/m<sup>3</sup>)
- **Oxygen below a Foundation: can it get there?**
  - Through the foundation
    - Cracks; concrete does have permeability to air
  - Around the foundation edges (bonus)
    - Oxygen has been found in sufficiently high quantities under most buildings
    - Large buildings or buildings built over dirty soils can be areas of low oxygen concentration

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# Dirty Soil Model for HC Vapors



# Vapor intrusion from direct building contact



## Preferential transport through a utility trench

